

SYNTHESIS OF ETHYL α -(DIETHYLPHOSPHONO)ACRYLATE AND ITS HOMOLOGS:
VERSATILE SYNTHETIC REAGENTS

Junya IDE,* Rokuro ENDO, and Shigeki MURAMATSU
Central Research Laboratories, Sankyo Co., Ltd.
1-2-58 Hiromachi, Shinagawa-ku, Tokyo 140

Ethyl α -(diethylphosphono)acrylate (2a) and ethyl β -methyl- α -(diethylphosphono)acrylates (E-2b and Z-2b) were prepared. The cycloaddition reaction of 2a with salicylaldehyde gave 3-ethoxycarbonyl-3-chromenes (9).

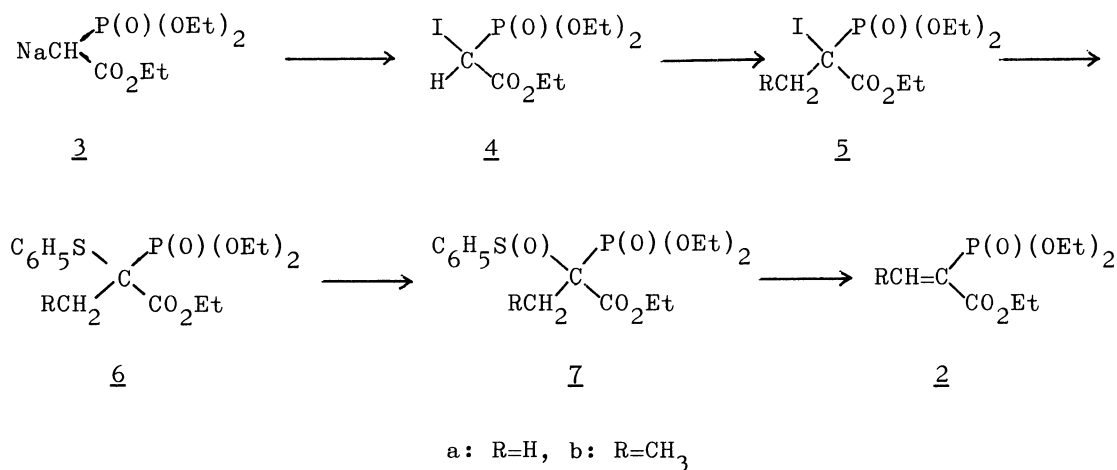
Vinyl triphenylphosphonium bromide (1) is a versatile synthetic reagent which supplies a two carbon unit required for the construction of heterocyclic and carbocyclic rings by Michael addition and subsequent intramolecular Wittig reaction.¹⁾ Ethyl α -(diethylphosphono)acrylate (2a) is expected to possess a functionality similar to 1, and the cycloaddition products derived from 2 (vide infra) provide an α,β -unsaturated carboxylate moiety as a functional group for their chemical transformation into more complex compounds.

Some ethyl α -(diethylphosphono)acrylates with a substituent at β -position were prepared from corresponding aldehydes and triethyl phosphonoacetate in the presence of titanium(IV) chloride by Lehnert.²⁾ Ethyl α -(diethylphosphono)acrylate (2a), however, could not be obtained according to Lehnert's method. We have found a facile method for the synthesis of 2a and ethyl β -methyl- α -(diethylphosphono)acrylate (2b).³⁾ Recent appearance of the literature⁴⁾ on 2a prompted us to report our results in a preliminary form.

Iodination of sodium triethyl phosphonoacetate (3) in dimethoxyethane (DME) with iodine (1 eq., 0°C, 10min.) gave triethyl α -iodophosphonoacetate (4), which, without isolation, was treated successively with sodium hydride (1.1 eq.,

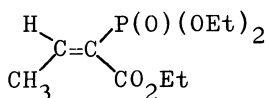
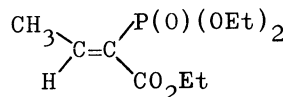
0°C, 15 min.) and methyl iodide (1.2 eq., 0°C, 30 min. then ambient temp., 1.5 hr) to afford the methylated phosphonate 5a [70% yield from 3, ν_{\max} (neat) cm^{-1} : 1730, 1240, 1045, 1020; δ (CDCl_3) ppm: 1.31(3H, t, $J=7$ Hz, $\text{CO}_2\text{CH}_2\text{CH}_3$), 1.38(6H, t, $J=7$ Hz, $\text{P}(\text{O})(\text{OCH}_2\text{CH}_3)_2$), 2.27(3H, d, $J_{\text{PCCH}}=15$ Hz, $^5\text{CH}_3$), 4.27(2H, q, $J=7$ Hz, $\text{CO}_2\text{CH}_2\text{CH}_3$), 4.20(4H, dq, $J_{\text{POCH}}=8$ and $J=7$ Hz, $\text{P}(\text{O})(\text{OCH}_2\text{CH}_3)_2$); m/e 364 (M^+)] .

Treatment of 5a with sodium thiophenolate (1.05 eq., 0°C to ambient temp., 1.5 hr) in DME gave the phenyl sulfide 6a [66% yield, ν_{\max} (neat) cm^{-1} : 1730, 1260, 1045, 1020; δ (CDCl_3) ppm: 1.23(3H, t, $J=7$ Hz, $\text{CO}_2\text{CH}_2\text{CH}_3$), 1.36(6H, t, $J=7$ Hz, $\text{P}(\text{O})(\text{OCH}_2\text{CH}_3)_2$), 1.56(3H, d, $J_{\text{PCCH}}=12$ Hz, CH_3), 4.25(6H, m, $\text{CO}_2\text{CH}_2\text{CH}_3$ and $\text{P}(\text{O})(\text{OCH}_2\text{CH}_3)_2$), 7.5(5H, m, $\text{C}_6\text{H}_5\text{S}$); m/e 346 (M^+)] . The phenyl sulfide 6a was oxidized with sodium periodate (1.2 eq.) in aq. methanol (ambient temp., 16 hr) to the corresponding sulfoxide 7a [97% yield, ν_{\max} (neat) cm^{-1} : 1740, 1260, 1050, 1025; m/e 362 (M^+)] . The sulfoxide 7a was heated under reflux in carbon tetrachloride to afford 2a [70% yield, ν_{\max} (neat) cm^{-1} : 1730, 1260, 1055, 1025; δ (CDCl_3) ppm: 1.33(9H, t, $J=7$ Hz, $\text{CO}_2\text{CH}_2\text{CH}_3$ and $\text{P}(\text{O})(\text{OCH}_2\text{CH}_3)_2$), 4.35(2H, q, $J=7$ Hz, $\text{CO}_2\text{CH}_2\text{CH}_3$), 4.22(4H, dq, $J_{\text{POCH}}=8$ and $J=7$ Hz, $\text{P}(\text{O})(\text{OCH}_2\text{CH}_3)_2$), 6.82(1H, dd, $J_{\text{PC=CH}}=22$ Hz and $J=2$ Hz, PC=CH (cis)), 5 7.07(1H, dd, $J_{\text{PC=CH}}=43$ Hz and $J=2$ Hz, PC=CH (trans)); m/e 236 (M^+)] .

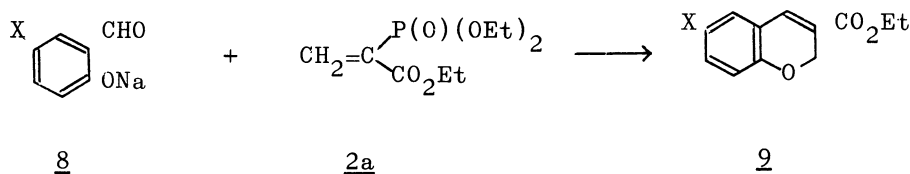


Ethyl β -methyl- α -(diethylphosphono)acrylate (2b) was similarly prepared as a mixture of the E- and Z-isomers, which were separated by silica-gel chromatography [the E-isomer E-2b; δ (CDCl_3) ppm: 1.33(9H, t, $J=7$ Hz, $\text{CO}_2\text{CH}_2\text{CH}_3$

and $P(O)(OCH_2CH_3)_2$, 2.11(3H, dd, $J=7$ Hz and $J_{PC=CH}=3$ Hz, CH_3), 4.28(6H, m, $CO_2CH_2CH_3$ and $P(O)(OCH_2CH_3)_2$), 7.30(1H, qd, $J_{PC=CH}=24$ Hz and $J=7$ Hz, $C=CH$), the Z-isomer Z-2b; 1.33(9H, t, $J=7$ Hz, $CO_2CH_2CH_3$ and $P(O)(OCH_2CH_3)_2$), 2.30(3H, dd, $J=7$ Hz and $J_{PC=CH}=3$ Hz, CH_3), 4.18(6H, m, $CO_2CH_2CH_3$ and $P(O)(OCH_2CH_3)_2$), 7.70(1H, qd, $J_{PC=CH}=46$ Hz and $J=7$ Hz, $C=CH$)].

E-2bZ-2b

As an example of the cycloaddition reactions, 2a was subjected to reaction with sodium salt of salicylaldehydes (8) to give 3-ethoxycarbonyl-3-chromenes (9) (Table).

Table. Formation of 9 by Cycloaddition Reaction of 2a with 8

X	Solvent	Reaction Temp.	Time(hr)	Yield(%)	m.p.(°C)
H	DMSO	r.t.	1	53	50-50.5
6-Cl	DMSO-THF	r.t.	1	42	118-119
6-Br	DMSO-THF	r.t.	2	19	124-125

Acknowledgement: We express appreciation to Dr. N. Soma, Director of Fine Chemical Research in our Laboratories, for his encouragement throughout this work.

References and Notes

- * To whom correspondence should be addressed.
- 1) E. E. Schweizer, J. Liehr and D. J. Monaco, J. Org. Chem., 33, 2416 (1968) and references cited therein; J. M. McIntosh and G. M. Masse, *ibid.*, 40, 1294 (1975) and references cited therein; W. G. Dauben and J. D. Robbins, Tetrahedron Lett., 151 (1975); I. Kawamoto, S. Muramatsu and Y. Yura, *ibid.*, 4223 (1974).
 - 2) W. Lehnert, Tetrahedron, 30, 301 (1974).
 - 3) Russian chemists reported the synthesis of 2a and 2b from the corresponding aldehydes and triethyl phosphonoacetate in the presence of piperidine in methanol under reflux: A. N. Pudovik, G. E. Yastrebova and V. I. Nikitina, Zhurnal Obshchei Khimii, 37, 2790 (1967), cf. C. A., 69, 43992s (1968). However, the desired compounds could not be obtained by their method.
 - 4) M. F. Semmelhack, J. C. Tomesch and A. Yamashita, Abstracts of papers: 26th International Congress of Pure and Applied Chemistry, Sept. 4-10, 1977, Tokyo, Session IV, p 1088; T. Minami, H. Suganuma and T. Agawa, Abstracts of papers: Symposium on Organic Sulfur and Phosphorous Compounds, Jan. 27-28, 1978, Osaka, p 65.
 - 5) G. Mavel in "Annual Reports on NMR Spectroscopy" Vol. 5B, E. F. Mooney, Ed., Academic Press, London and New York, 1973, p 49; G. Mavel in "Progress in NMR Spectroscopy" Vol. 1, J. W. Emsley, J. Feeney and L. H. Sutcliffe, Ed., Pergamon Press, 1966, p 261.

(Received January 25, 1978)